Occupational Exposure Assessment of Organic Pollutants Emitted Through Aerosols from Recycling of e-Waste

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Abstract
The emission of airborne particulate matter is produced by the industrial processes of metal extraction from electronic waste (e-waste). These aerosols can transport organic contaminants, such as polycyclic aromatic hydrocarbons and polybrominated biphenyls. These pollutants are proven to have long-term effects on workers’ health. In this work, we characterize aerosol particles and dust deposited from a recycling process of printed circuit boards and waste cables in a recycling plant. A Scanning Mobility Particles Counter and Sizer (SMPCS) was used to measure the size distributions of ultrafine particles. A Micro-Orifice Uniform Deposition Impactor (MOUDI) collected aerosol particle samples for chemical analysis. All plant compartments checked are subject to high particulate matter concentrations during work hours. Concentrations of organic pollutants in the aerosols do not pose a short, or long-term health risk. However, the accumulation of aerosols in other compartments in the form of dust may present an indirect danger to workers’ health.

Keywords: WEEE; Recycling; aerosol; PBDEs; PAHs; Exposure
Introduction

Electronic scrap also called e-waste or waste electrical and electronic equipment (WEEE) is an emerging part of municipal solid waste [1,2]. Recycling of WEEE bares the advantage of the high commercial value of the end products (metals) [3,4]. Toxic substances albeit in small amounts could become a serious concern to the environment and the personnel of a recycling plant [5].

In e-waste recycling, brominated flame retardants (BFRs), from Printed Circuit Boards (PCBs), are perhaps the most harmful organic compounds [6,7]. BFRs are the bromodioxin major contributors since they can produce halogenated (brominous) dioxins and furans. They are very often used in WEEE electronic circuits, sometimes up to 20 percent wt [8,9]. BFR use in 2000 was 32,000 tn in Europe and around 540,000 tn worldwide [10]. Some of the most common BFRs are Tetrabromobisphenol A (TBBP-A), Polybrominated Biphenyls (PBB), Polybrominated diphenyl ethers (PBDE), Hexabromocyclododecane (HBCD), Phenylethers and Brominated Bisphenol [11]. BFRs can generate brominated dioxins through chemical reduction at temperatures from 350 to 400°C. Bromodioxins can even be randomly produced [12,13] or even from decabromodiphenyl ether after exposure to UV rays [14]. The nature of the processes of WEEE recycling, such as milling, shredding, and grinding of PCBs and cables, as well as catalytic influences of certain substances (Sb2O3, Fe2O3,) and also moisture, enhance the generation of bromodioxins [15-20]. Bromodioxins could be transported by particulate matter with dimensions ranging from several nm to tens of micrometers.

Aerosols themselves (i.e. particles having a size lower than 100 nanometres) are a very common health risk in industrial plants [21,22]. A growing amount of research proved that pulmonary and heart diseases are directly associated with aerosols themselves [23-26]. The association of processes between the human organism and aerosols has still not been properly understood [27]. Nevertheless, a significant amount of toxicological and epidemiological research indicated that smaller aerosols have more acute health consequences [28-30].

In this study, we document measurements of aerosols produced by printed circuit boards (PCBs) and waste cables milling in a small-scale recycling industry and measure BFRs and polyaromatic hydrocarbons (PAHs) in airborne particulate matter. An evaluation of an employee’s inhalation absorption is also implemented to assess the employee’s protection level from the particulate matter, and the organic contaminants carried with it.

Materials and Methods

Description of the Facility-Measurement locations

All analyses were carried out at a recycling facility that serves the needs of a population of 85,000. The firm’s operations include plastics and paper packaging, e-waste, discarded copper wires, discarded metals, and waste automobiles. The factory is thus split between different sectors assigned to different recycling activities.

The factory’s building consists of two areas (Figure 1, top), segregated by a wall. There is no intermediate communication between the two areas, and thus, no direct internal influence between them. PCBs and waste cables recycling equipment is located in the eastern area of the ground level. The equipment includes machinery for shredding, grinding sieving, and separation with air cyclones, electric sieves, and vibrating tables. All these series of equipment comprise two individual systems of PCBs and cable recycling. Their function is to distinguish the metallic and non-metallic fractions of the cables and PCBs. This is the first location of examination and the contamination source (Figure 1, A, Measurement Location A, MLA). The non-metallic fraction of the PCBs is made of compressed cardboard, woven glass and epoxy resin, phenolic cotton paper, cotton paper, and epoxy resin, and matt and polyester glass [31].
The PCB's non-metallic parts and the ground plastic insulation of the cables are guided by a common airflow outside the building and are collected in bags. The room west of the aforementioned territory is allocated to steel processing and welding odd jobs, repairs, and large household appliances manual dismantling (washing machines, refrigerators, cookers). This is the second location of examination (Figure 1, B, Measurement Location B, MLB).

The other area of examination (Figure 1 C, Measurement Location C, MLC) is a distant location outside the factory buildings, situated in the factory's southwest part. This area used to host a cathode ray tube processing area, however, during the time of measurements, the recycling line had been relocated to other parts of the plant. This location is situated behind a pile of scrap metals as seen in Figure 1. Thus, it remained to be determined whether it was possible for this location to be influenced by the indirect release of pollutants from the

**Figure 1**: Schematic diagram of the site and the particulate matter measurement locations at (A-MLA) the printed circuit boards and cables processing territory, (B-MLB) the workshop and large appliances dismantling line, and (C-MLC) remote point outside the plant.
source of contamination, the PCBs and cables grinding and processing location.

**Measurements of Airborne Particulate Matter**

A Grimm Model 5416 Scanning Mobility Particle Counter (SMPCS) monitored the aerosol dimension distributions at the aforementioned factory areas. The SMPCS monitored the distribution of the mobility diameter of the aerosols from dimensions of 5 to ca 1094 nm. The inlet flow rate of the SMPCS was 0.3 L/min, and the measurement period was 3 min [32-34]. Aerosols that were greater than ca. 1094 nm were excluded from the air current circuit. The timeframe required to calculate the diameter distributions of aerosols with dimensions from 10 to ca 1094 nm in the measurements was 3 min. Aerosol Instrument Manager TSI version 6.0 monitored the raw data.

Additionally, a multi stage cascade impactor (Thermo Scientific 20-830-SS Marple and Cascade Impactor) was used, in order to capture aerosols for PBDEs chemical examination. The filters used were 0.47 cm high-purity quartz (SiO$_2$) microfiber filters (Whatman Qm-A). The impactor collected aerosols using 2 chambers of impaction for aerosols with dimensions less than 1 μm (PM1) and less than 10 μm (PM10), in accordance with the methodology proposed by Marple et al. [35]. The pump operated at approximately 24 m$^3$ per day.

Particulate matter mass and number concentrations were measured by the cascade impactor and the SMPCS simultaneously, by sampling ambient air by a common inlet tube traversing a dehumidifier and divided into two streams for each instrument, as depicted above (Figure 2).

**Figure 2:** The configuration of the cascade impactor and the scanning mobility particle sizer (SMPS)

The inlet tube was a 60x8 cm tubular humidity-reducing configuration packed with a yellow bead from silica gel (Silikagel Trockenperlen, Azelis GmbH).

Over the warmer months of the year (end of August), aerosol data and samples were collected out for 6 days in the three aforementioned discrete locations around the recycling center facilities (as seen in Figure 1).
Analysis of Polyaromatic Hydrocarbons (PAHs) and Polybrominated Diphenyl Ethers (PBDEs) in the Particle Samples Particulate Matter

Analysis of PBDEs and PAHs – Liquid-liquid extraction method

For PBDEs analysis, collected substances from the quartz filters of the impactor were extracted with dichloromethane: hexane (1:1) in an accelerated solvent extraction system. Prior to extraction, the quartz filters were spiked with a range of recovery standards for PBDEs.

The extract was rotary evaporated to 1 ml and treated with concentrated H$_2$SO$_4$ until no color was visible in the acidic phase. Then the organic layer was separated from H$_2$SO$_4$, evaporated to 0.5 mL, and transferred to a Pasteur pipette column packed with silica gel (SiO$_2$) in the bottom (1 cm) and 40% sulphuric acid silica gel (40% H$_2$SO$_4$–SiO$_2$) at the top (3 cm). The sample was eluted with 8 mL of dichloromethane and the volume was reduced to 0.5 ml. The solvent was exchanged with hexane and the sample was applied to a Pasteur pipette column packed with silica gel activated at 300°C for 3 h. The column was eluted with hexane (first fraction) and dichloromethane in hexane (second fraction). Subsequently, the second fraction was rotary evaporated to 200 μL, transferred into a glass vial, and further evaporated to approximately 5 μL under a stream of N$_2$. Finally, a BDE standard solution was added as an internal standard, to achieve a final volume of 25 μL.

For PAH analysis, substances from the filters (about 0.5 g from each) were Soxhlet extracted with toluene for 24 h and the extracts were evaporated to almost dryness. Quantities of half a microgram of polyaromatic hydrocarbons doped with D$_3$ were added to each extract. N – C$_6$H$_{12}$ was used as the exchange agent for the solution and the optimized liquid–liquid technique took place. 5% dimethylformamide ((CH$_3$)$_2$NC(O)H) solutions in ultra-pure water were used to part 2 ml of the n – C$_6$H$_{12}$ solution twice. Both (CH$_3$)$_2$NC(O)H partitions were added together in a beaker. Subsequently, they were transferred in glass tubes. A quantity of 4 ml of C$_6$H$_{12}$ solution was then used to retrograde extract the polyaromatics with the addition of another 4 ml of ultra-pure H$_2$O. In continuation, a repetition of the back extraction took place with another 4 ml of C$_6$H$_{12}$. The C$_6$H$_{12}$ quantity was then decreased. The chemical efficiency of the technique was higher than 80, and up to 100%.

Extracts of PAHs and PBDEs from particulate matter were analyzed by a gas chromatograph (Agilent 6890 Series, Agilent Technologies, Palo Alto, CA, USA), which consists of a cool on-column injector combined with a mass spectrometer (Agilent 6890 Series, Agilent Technologies, Palo Alto, CA, USA) based on selected ion monitoring using electron impact ionization (70 eV electron energy).

Analytes were separated on a 15-m DB5-MS capillary column operating with a helium carrier gas of constant velocity 46 cm/s with 5 phenyl- methylpolysiloxane, 0.25 nm internal diameter, 0.1 lm film thickness. The following temperature program was used: from 90°C to 140°C at 50°C/min, to 220°C at 15°C/min and finally to 295°C at 10°C/min. The temperature of the injector was initially set at 93°C and then followed the temperature program of the oven. The temperature of the transfer line, ion source, and quadrupole filter was kept constant at 300°C, 230°C, and 150°C, respectively.

Discussion

SMPCS Measurements of Aerosols

Contour plots of SMPCS measurements are shown in figures 4, 5, and 6. Contour plots at the contamination source (M-LA), shown in figure 4, manifest the plant activity which occurred constantly between 07:00 till 16:00. Plots at the dismantling workshop (MLB), shown in figure 5, indicate that the influence of the cable and PCB processing room is marginally evident. The two rooms are separated by a wall; thus, any influence would originate from external ambient air coming into the room rather than a direct influx between rooms.
Figure 4: Contour plots of SMPS measurements over the first and second day, at the circuit boards and cables processing territory (MLA), and b) Total, nucleation, accumulation, and Aitken aerosols. c, d) Charts of distribution of particles in two instances: First, during operation of the processing machines of PCBs (left, territory marked as “a” in the previous graph), and secondly and secondly during shutdown (right territory marked as “b” in the previous graph).

Contour plots over the last two days at the blind spot (MLC) are shown in Figure 6. The disturbance in the high concentrations of ultra-fine particles is evident. This shows the influence of the contamination source on the external plant conditions. The density of the nanoparticles in that spot (nucleation fraction) seems to increase more than their larger particle counterparts (i.e., accumulation and Aitken particles), indicating that smaller particles travel longer distances.

Organic Substances. PBDEs and PAHs

The results of chemical analysis for PBDEs and PAHs in the aerosols are shown on Figures 7 and 8. The total concentration of PAHs is on a scale of 130 to 350 pg/m^3. In PAH analysis phenanthrene, a polycyclic aromatic hydrocarbon composed of three fused benzene rings has the highest concentration of all PAH species in all measurement locations (38 to 75 pg/m^3). Pyrene, a PAH consisting of four fused benzene rings, resulting in a flat aromatic system with chemical formula C_{16}H_{10} is also dominant (6 to 50 pg/m^3). Fluorene, or 9H-fluorene, is a polycyclic aromatic hydrocarbon that also appears in notable quantities (8 to 53 pg/m^3). Benzo(ghi)perylene a 6-ring polycyclic aromatic hydrocarbon with the chemical formula C_{22}H_{12} appears in high concentrations at the remote point aerosols. Benzopyrene, with the formula C_{20}H_{12}, a pentacyclic hydrocarbon with its isomeric species, benzo(a)pyrene and benzo(e)pyrene reaches 23 pg/m^3. The concentration of PAH pollutants in PM10 particles is generally higher than PM1 in all compartments checked.
The sources of PAHs can be attributed to vehicles passing the neighboring road, the milling processes of the PCBs and cables, and the operation of heavy-duty vehicles with machine tools in the yard, that use diesel fuel.

The total concentration of PBDEs ranges from 30 to 145 ng/m$^3$. PBDE 47 and PBDE 99 have the highest concentration, ranging from 12 to around 60 ng/m$^3$. The concentration of PBDE pollutants in PM10 particles is also higher than PM1 in all locations. The highest concentrations are observed at the dismantling workshop, therefore aerosols rich in PBDEs accumulate in the workshop. Thus, despite the fact that the contamination source does not influence directly the dismantling workshop, the accumulation and resuspension of aerosols increase the pollution burden of the territory.

Another conclusion of the above analysis is that polybrominated compounds dominate the organic content of the aerosols compared to PAHs by a factor of 300/1 or more. PBDEs permeate into the aerosol phase through the dust that is created during the pulverization of the printed circuit boards. Pulverization, due to the high temperatures, is assumed to assist in the passing of the PBDEs to the airborne phase.

PDBEs may be either formed and emitted directly as small aerosol particles (chunks of recycled materials that have a diameter of the order of microns or smaller) or as gases that then condense on the emitted particles.
Exceptionally high concentrations appear for PBDE 47\(\text{2,2',4,4'-tetrabromodiphenyl ether}\) and PBDE 99\(\text{2,2',4,4',5-pentabromodiphenylether}\) that reach almost 60ng/m^3 in the aerosols collected.

The areas mostly affected by this source of contamination are primarily the dismantling workshop and the contamination source. Despite the contamination source has the source of PBDEs contamination, small amounts of the emitted particles are transported to the neighboring area of the workshop. The transported aerosols are exceptionally rich in PAHs and PBDEs. Therefore, the overall PBDEs and PAHs contamination of the dismantling workshop is higher than in the other two territories.

The aerosols could reach the blind spots and therefore the organic contaminants could be transferred further away from the source of contamination with the help of the aerosols. For example, phenanthrene for the PM10 fraction at the blind spot reached almost 63 pg/m^3 in the aerosols.

In a previous study \[36\] of PBDEs concentrations in offices and internet, cafés were on a scale of 100 to 200 pg/m^3, significantly lower than this study, whereas PBDEs in urban areas in another work \[37\] were found around 4 to 45 pg/m^3. These values are also well above the values measured by Deng et al. \[38\] (21.5 ng/m^3 in total suspended particulate matter). Conclusively, PBDE concentrations in the scale of ng/m^3, are orders of magnitude higher compared to concentrations found in other e-waste recycling workplaces or in general in the environment.
Figure 7: PAH analysis of PM1 and PM10 filters of three days, one for each region
PAHs and PBDEs Worker Absorption Evaluation

Construction of Model

The doses received by air contaminants via factory practices rely on a variety of variables, such as the features of the compartments, the cycle of operations, the weather [39], the density of the aerosol contaminants within a dormitory, their fluctuation from day to day or even between hours [40,41], the density of aerosols (which is inversely proportional to the range from the contamination center, the existence of narrow paths and barriers inside a room - even actions of humans [42]. Attention must also be attributed to the existence and treatment of powder materials, even non-airborne dust, their dimensions, moisture, static electricity, charge [43-45], even heat transfers from the human body, and turbulences created by human limp movements [46-48]. Computational fluid dynamics (CFD) models may be effective regarding all the aforementioned, however, they produce wide errors and non-predictable results [49] as only abstract predictions regarding their impact are possible, with wide range of error [50]. Thus, in this work, these con-factors will not be regarded individually and the aerosol density will be assumed to be a continuum gradient in each independent territory [51].

A determining factor of a human’s hazardous substances intake through the lungs is the respiration rate. This is the quantity of air breathed by an average adult human within a predetermined time interval [52,53]. Researchers were able to estimate the respiration rate through the quantity of oxygen retained by the human body in comparison to the air quantity inhaled [54-56]. Age and sex patterns and four types of work intensity were also regarded in our work: Inactive, light activity, medium activity, and intense activity. Most types of tasks undertaken in the factory are executed by the use of automated machinery and tools. Direct manual tasks are very rare and occur arbitrarily during the work cycle. Thus, it is secure for the calculations to assume the work intensity of light activity.

Organic Contaminants Intake Estimation

According to the calculations of EPA described before and by considering work intensity of light activity the mean volume of air intake during the entire time length of the work shift for men and women workers, and for several categories from 20 to 70 years is listed in Table 2. A more detailed calculation of the constructed model is described by Lasithiotakis et. al. 2019 [57].

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Table 2: Average air intake during a workday for men and women, and for age categories of 21 to 71 years

<table>
<thead>
<tr>
<th>Age Group (Years)</th>
<th>Average Air Breathen (m3/8h shift)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Mean</td>
</tr>
<tr>
<td>21 to 31</td>
<td>14</td>
</tr>
<tr>
<td>31 to 41</td>
<td>14.5</td>
</tr>
<tr>
<td>41 to 51</td>
<td>15.2</td>
</tr>
</tbody>
</table>
By these assessments, minimum and maximum intake calculations can take place so as to estimate organic contaminants intake by an individual in PAHs and PBDEs of the particulate matter. Total absorption of particulate matter in the alveoli of the lungs is considered. The maximum values hypothesis developed in this work estimates the mean worker exposure within a day of PAHs and PBDEs, for the highest amount of air inhaled within a workday. This is 34.4 cubic meters for men in the group 20 to 30 years old. The above hypothesis regards that the individual should remain and be active in the aforementioned areas during the whole time of their shift, where the maximum concentrations of organic contaminants were observed, and for the particulate matter of range sizes of 1μm or less. A list of these results, for PBDEs and PAHs, is displayed in Tables 3 and 4.

Table 3: Calculation of PAHs, absorption during a day work shift for men of 20 to 30 years of age

<table>
<thead>
<tr>
<th>Nanoparticle Size</th>
<th>MLA PCBS and cables Recycling Area</th>
<th>MLB Workshop and Dismantling</th>
<th>MLC Blind Spot Recycling Area</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>PM1</td>
<td>PM10</td>
<td>PM1</td>
</tr>
<tr>
<td>Intake (ng/Sh shift)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fluorene</td>
<td>4.6E-01</td>
<td>4.6E-01</td>
<td>2.9E-01</td>
</tr>
<tr>
<td>Phenanthrene</td>
<td>2.2E+00</td>
<td>1.9E+00</td>
<td>2.0E+00</td>
</tr>
<tr>
<td>Anthracene</td>
<td>1.7E-01</td>
<td>2.2E-01</td>
<td>1.1E-01</td>
</tr>
<tr>
<td>C1-Phe</td>
<td>2.7E-01</td>
<td>2.4E-01</td>
<td>2.0E-01</td>
</tr>
<tr>
<td>C2-Phe</td>
<td>3.7E-01</td>
<td>3.2E-01</td>
<td>2.4E-01</td>
</tr>
<tr>
<td>C3-Phe</td>
<td>5.2E-01</td>
<td>3.4E-01</td>
<td>2.9E-01</td>
</tr>
<tr>
<td>C4-Phe</td>
<td>2.4E-01</td>
<td>1.8E-01</td>
<td>2.0E-01</td>
</tr>
<tr>
<td>3.6 DMP</td>
<td>1.0E-01</td>
<td>6.2E-02</td>
<td>4.4E-02</td>
</tr>
<tr>
<td>2.6 DMP</td>
<td>5.6E-02</td>
<td>4.9E-02</td>
<td>2.0E-02</td>
</tr>
<tr>
<td>2.7 DMP</td>
<td>6.8E-02</td>
<td>7.2E-02</td>
<td>3.9E-02</td>
</tr>
<tr>
<td>1.3/2.10/3.9/3.10 DMP</td>
<td>4.0E-02</td>
<td>4.3E-02</td>
<td>9.8E-03</td>
</tr>
<tr>
<td>1.6/2.9 DMP</td>
<td>1.7E-01</td>
<td>1.6E-01</td>
<td>7.5E-02</td>
</tr>
<tr>
<td>Fluoranthene</td>
<td>7.2E-01</td>
<td>4.7E-01</td>
<td>2.6E-01</td>
</tr>
<tr>
<td>1.7 DMP</td>
<td>8.4E-02</td>
<td>7.1E-02</td>
<td>3.1E-02</td>
</tr>
<tr>
<td>2.3 DMP</td>
<td>1.1E-01</td>
<td>9.1E-02</td>
<td>1.1E-01</td>
</tr>
<tr>
<td>1.9/4.9 DMP</td>
<td>7.6E-02</td>
<td>6.9E-02</td>
<td>4.4E-02</td>
</tr>
<tr>
<td>1.8 DMP</td>
<td>2.0E-02</td>
<td>2.9E-02</td>
<td>1.3E-02</td>
</tr>
<tr>
<td>Substance</td>
<td>PM1</td>
<td>PM10</td>
<td>PM1</td>
</tr>
<tr>
<td>--------------------</td>
<td>------</td>
<td>------</td>
<td>------</td>
</tr>
<tr>
<td>Pyrene</td>
<td>1.7E+00</td>
<td>6.2E-01</td>
<td>2.2E-01</td>
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<tr>
<td>C1-202</td>
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<td>4.8E-02</td>
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<td>3.9E-03</td>
</tr>
<tr>
<td>Ret</td>
<td>0.0E+00</td>
<td>0.0E+00</td>
<td>0.0E+00</td>
</tr>
<tr>
<td>4H-Cy</td>
<td>1.4E-01</td>
<td>9.0E-02</td>
<td>1.8E-02</td>
</tr>
<tr>
<td>Benzo(a)anthracene</td>
<td>5.5E-02</td>
<td>2.5E-01</td>
<td>2.7E-02</td>
</tr>
<tr>
<td>Chrysene</td>
<td>1.8E-01</td>
<td>4.6E-01</td>
<td>6.0E-02</td>
</tr>
<tr>
<td>C1-Crysene</td>
<td>3.3E-02</td>
<td>1.2E-01</td>
<td>0.0E+00</td>
</tr>
<tr>
<td>Benzo(b)fluoranthene</td>
<td>1.9E-01</td>
<td>4.7E-01</td>
<td>6.5E-02</td>
</tr>
<tr>
<td>Benzo(k)fluoranthene</td>
<td>6.5E-02</td>
<td>2.7E-01</td>
<td>5.0E-02</td>
</tr>
<tr>
<td>Benzo(bj)fluoranthene</td>
<td>7.2E-03</td>
<td>5.1E-02</td>
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<tr>
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<td>4.1E-01</td>
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<tr>
<td>Benzo(a)pyrene</td>
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<td>4.0E-02</td>
<td>4.3E-03</td>
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<tr>
<td>Anthranene</td>
<td>3.2E-02</td>
<td>1.0E-01</td>
<td>3.2E-02</td>
</tr>
<tr>
<td>Indeno(123cd)pyrene</td>
<td>1.2E-01</td>
<td>3.2E-01</td>
<td>7.9E-02</td>
</tr>
<tr>
<td>Dibenzo(ah)anthracene</td>
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<td>8.0E-02</td>
<td>0.0E+00</td>
</tr>
<tr>
<td>Benzo(ghi)perylen</td>
<td>e</td>
<td>1.3E-01</td>
<td>4.9E-01</td>
</tr>
<tr>
<td>Coronene</td>
<td>0.0E+00</td>
<td>0.0E+00</td>
<td>0.0E+00</td>
</tr>
<tr>
<td>Sum</td>
<td>8.8</td>
<td>9</td>
<td>4.7</td>
</tr>
</tbody>
</table>

**Table 4:** Calculation of or PBDEs absorption during a day work shift for men of 20 to 30 years of age
Table 4 shows that the above results are generally below the aforementioned recommended limits. However, this does not annul the possibility for the contaminants to have synergistic effects on workers’ health. Therefore, personal protective equipment during work and good maintenance of ventilation apparatuses are recommended.

In this hypothetical case, the number of PAHs inhaled and absorbed through particulate matter within a shift is 9.1 ng, for PM10 at MLB (dismantling workshop), in the time frame of one workday is 9.1 nanograms, for PM 10 and the PBDEs around 5μg at the same location.

According to US Occupational Safety and Health Administration (OHSA), [58] the permissible exposure level for PAHs in the workplace for an 8-h time-weighted average is 200 ng/m3, which is well above the highest concentrations measured in this work. For PBDEs, there is no internationally accepted limit of intake yet. A value of 5 mg/m3 for deca-BDE, with ongoing air monitoring, has been proposed as well as 6 μg/m3 for lower brominated PBDEs [59,60] both above the values measured.

Conclusions

Particulate matter contamination could move from compartment to compartment accumulating in other areas that also resuspended fine particles. The dust was produced at the mills and granulators of the recycling lines of the printed circuit boards and cables. However, the dismantling workshop exhibited higher concentrations of aerosol organic pollutants than the recycling line. Polybrominated compounds dominate the organic content of the aerosols compared to PAHs. PBDEs can permeate and transport to longer distances into the aerosol phase through the dust. Other areas can be affected by this source of contamination such as the workshop and the blind spot. The outside blind spot exhibited lower contamination. Nevertheless, all the above results are generally below the recommended limits. However, transportation, accumulation, and resuspension of contaminants through aerosols could aggravate occupational health conditions.
References


58. Occupational Safety & Health Administration (OSHA) Coal Tar Pitch Volatiles
